

WEST Search History

DATE: Saturday, March 31, 2007

<u>Hide?</u>	<u>Set</u>	<u>Name</u>	<u>Query</u>	<u>Hit Count</u>
<i>DB=PGPB,USPT,USOC,EPAB,JPAB,DWPI; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>				
<input type="checkbox"/>	L4		L2 and separat\$3 near4 carbon dioxide	5
<input type="checkbox"/>	L3		L2 and separat\$3 near4 carbon dioxide near4 (synthesis gas or syngas)	0
<input type="checkbox"/>	L2		L1 and combust\$3 with oxygen same catalyst	15
<input type="checkbox"/>	L1		(hydrocarbon feedstock or natural gas or methane) and cataly\$2 near4 steam same heat exchange reform\$3	28

END OF SEARCH HISTORY

Hit List

First Hit	Clear	Generate Collection	Print	Fwd Refs	Bkwd Refs
Generate OACS					

Search Results - Record(s) 1 through 5 of 5 returned.

1. Document ID: US 20070010590 A1

L4: Entry 1 of 5

File: PGPB

Jan 11, 2007

PGPUB-DOCUMENT-NUMBER: 20070010590

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20070010590 A1

TITLE: Production of hydrocarbons by steam reforming and fischer-tropsch reaction

PUBLICATION-DATE: January 11, 2007

INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY
Abbott; Peter Edward James	Cleveland		GB
McKenna; Mark	Stockton on Tees		GB

APPL-NO: 10/555369 [PALM]

DATE FILED: April 21, 2004

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	DOC-ID	APPL-DATE
GB	0310106.0	2003GB-0310106.0	May 2, 2003

PCT-DATA:

DATE-FILED	APPL-NO	PUB-NO	PUB-DATE	371-DATE
Apr 21, 2004	PCT/GB04/01677			Jul 20, 2006

INT-CL-PUBLISHED:

TYPE	IPC	DATE	IPC-OLD
IPCP	C07C27/06	20060101	C07C027/06

INT-CL-CURRENT:

TYPE	IPC	DATE
CIPP	<u>C07 C 27/06</u>	20060101

US-CL-PUBLISHED: 518/703

US-CL-CURRENT: 518/703

ABSTRACT:

A process for the production of hydrocarbons is described including; a) subjecting a mixture of a hydrocarbon feedstock and steam to catalytic steam reforming to form a partially reformed gas, b) subjecting the partially reformed gas to partial combustion with an oxygen-containing gas and bringing the resultant partially combusted gas towards equilibrium over a steam reforming catalyst to form a reformed gas mixture, c) cooling the reformed gas mixture to below the dew point of the steam therein to condense water and separating condensed water to give a de-watered synthesis gas, d) synthesising hydrocarbons from side de-watered synthesis gas by the Fischer-Tropsch reaction and e) separating the hydrocarbons from co-produced water, characterised in that at least part of said co-produced water is fed to a saturator wherein it is contacted with hydrocarbon feedstock to provide at least part of the mixture of hydrocarbon feedstock and steam subjected to steam reforming.

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KMC	Drawn
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2. Document ID: US 20060135629 A1

L4: Entry 2 of 5

File: PGPB

Jun 22, 2006

PGPUB-DOCUMENT-NUMBER: 20060135629

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20060135629 A1

TITLE: Production of hydrocarbons

PUBLICATION-DATE: June 22, 2006

INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY
Abbott; Peter Edward James	Cleveland		GB
Fowles; Martin	North Yorkshire		GB

APPL-NO: 10/534239 [PALM]

DATE FILED: October 23, 2003

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	DOC-ID	APPL-DATE
GB	0225961.2	2002GB-0225961.2	November 7, 2002

PCT-DATA:

DATE-FILED	APPL-NO	PUB-NO	PUB-DATE	371-DATE
Oct 23, 2003	PCT/GB03/04622			Nov 8, 2005

INT-CL-PUBLISHED:

TYPE	IPC	DATE	IPC-OLD
	IPCP C07C27/06	20060101	C07C027/06

INT-CL-CURRENT:

TYPE	IPC	DATE
	CIPP C07 C 27/06	20060101

US-CL-PUBLISHED: 518/702
US-CL-CURRENT: 518/702

ABSTRACT:

A process for production of hydrocarbons including a) reforming a divided hydrocarbon feedstock stream, mixing the first stream with steam, passing the mixture over a catalyst disposed in heated heat exchange reformer tubes to form a primary reformed gas, forming a secondary reformer feed stream including the primary reformed gas and the second hydrocarbon stream, partially combusting the secondary reformer feed stream and bringing the partially combusted gas towards equilibrium over a secondary catalyst, and producing a partially cooled reformed gas, b) further cooling the partially cooled reformed gas below the dew point of steam therein to condense water and separating condensed water to give a de-watered synthesis gas, c) synthesising hydrocarbons from the de-watered synthesis gas by the Fischer-Tropsch reaction and separating some of the synthesised hydrocarbons into a tail gas, and d) incorporating part of the tail gas into the secondary reformer feed stream before partial combustion thereof.

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Drawn De

3. Document ID: US 20020006968 A1

L4: Entry 3 of 5

File: PGPB

Jan 17, 2002

PGPUB-DOCUMENT-NUMBER: 20020006968
PGPUB-FILING-TYPE: new
DOCUMENT-IDENTIFIER: US 20020006968 A1

TITLE: Steam reforming

PUBLICATION-DATE: January 17, 2002

INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY
Abbott, Peter Edward James	Eaglescliffe		GB

APPL-NO: 09/781312 [PALM]
DATE FILED: February 13, 2001

RELATED-US-APPL-DATA:
child 09781312 A1 20010213
parent continuation-of PCT/GB99/02286 19990715 US UNKNOWN

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	DOC-ID	APPL-DATE
GB	9817526.8	1998GB-9817526.8	August 13, 1998

INT-CL-PUBLISHED: [07] C07C 27/06

INT-CL-CURRENT:

TYPE	IPC	DATE
CIPS	<u>C01 B 3/00</u>	20060101
CIPS	<u>C10 G 2/00</u>	20060101
CIPS	<u>B01 J 19/00</u>	20060101
CIPS	<u>B01 J 8/02</u>	20060101
CIPS	<u>B01 J 8/04</u>	20060101
CIPS	<u>B01 J 8/06</u>	20060101
CIPS	<u>C01 B 3/38</u>	20060101

US-CL-PUBLISHED: 518/704

US-CL-CURRENT: 518/704

REPRESENTATIVE-FIGURES: NONE

ABSTRACT:

Production of synthesis gas for use for synthesising carbon-containing compounds, typically having a hydrogen to carbon monoxide molar ratio of about 2 and a low carbon dioxide content, by primary reforming a gaseous mixture containing hydrocarbons, 0.6 to 2 moles of steam per gram atom of hydrocarbon and 0.2 to 0.6 moles of recycled carbon dioxide per gram atom of hydrocarbon, in a heat exchange reformer and then secondary reforming the resultant primary reformed gas, heating the heat exchange reformer with the resultant secondary reformed gas; cooling and condensing steam from the secondary reformed gas to give a de-watered gas stream having a carbon dioxide content below 20% by volume. The recycled carbon dioxide is recovered from the de-watered gas stream, before or after use thereof for the synthesis reaction.

[Full](#) | [Title](#) | [Citation](#) | [Front](#) | [Review](#) | [Classification](#) | [Date](#) | [Reference](#) | [Sequences](#) | [Attachments](#) | [Claims](#) | [KMC](#) | [Draw](#) [D](#)

4. Document ID: US 7097925 B2

L4: Entry 4 of 5

File: USPT

Aug 29, 2006

US-PAT-NO: 7097925

DOCUMENT-IDENTIFIER: US 7097925 B2

TITLE: High temperature fuel cell power plant

DATE-ISSUED: August 29, 2006

PRIOR-PUBLICATION:

DOC-ID	DATE
US 20030143448 A1	July 31, 2003

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Keefer; Bowie G.	Galiano Island			CA

ASSIGNEE-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY	TYPE CODE
QuestAir Technologies Inc.	Burnaby			CA	03

APPL-NO: 10/352361 [PALM]
 DATE FILED: January 27, 2003

RELATED-US-APPL-DATA:

continuation-in-part parent-doc US 10039940 00 20011026 PENDING child-doc US
 10352361
 us-provisional-application US 60351798 00 20020125
 us-provisional-application US 60323169 00 20010917

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	APPL-DATE
CA	2325072	October 30, 2000

INT-CL-ISSUED:

TYPE	IPC	DATE	IPC-OLD
IPCP	H10M8/12	20060101	H10M008/12
IPCS	H01M8/24	20060101	H01M008/24

INT-CL-CURRENT:

TYPE	IPC	DATE
CIPP	<u>H01 M 8/12</u>	20060101
CIPS	<u>H01 M 8/24</u>	20060101

US-CL-ISSUED: 429/9; 429/30, 429/34

US-CL-CURRENT: 429/9; 429/30, 429/34

FIELD-OF-CLASSIFICATION-SEARCH: 429/13, 429/17, 429/19, 429/20, 429/30, 429/31, 429/34, 429/9

See application file for complete search history.

PRIOR-ART-DISCLOSED:

U.S. PATENT DOCUMENTS

PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<u>3094569</u>	June 1963	Thomas	
<u>3204388</u>	September 1965	Asker	
<u>3430418</u>	March 1969	Wagner	
<u>3513631</u>	May 1970	Siebert et al.	
<u>3564816</u>	February 1971	Batta	
<u>3594984</u>	July 1971	Toyama et al.	
<u>3847672</u>	November 1974	Trocciola et al.	
<u>3865924</u>	February 1975	Gidaspow et al.	
<u>4019879</u>	April 1977	Rabo et al.	
<u>4144037</u>	March 1979	Armond et al.	
<u>4153434</u>	May 1979	Settlemeyer	

<u>4200682</u>	April 1980	Sederquist
<u>4272265</u>	June 1981	Snyder
<u>4322394</u>	March 1982	Mezey et al.
<u>4354859</u>	October 1982	Keller et al.
<u>4406675</u>	September 1983	Dangieri et al.
<u>4452612</u>	June 1984	Mattia
<u>4530705</u>	July 1985	Firey
<u>4532192</u>	July 1985	Baker et al.
<u>4553981</u>	November 1985	Fuderer
<u>4555453</u>	November 1985	Appleby
<u>4578214</u>	March 1986	Jungerhans
<u>4587114</u>	May 1986	Hirai et al.
<u>4595642</u>	June 1986	Nakanishi et al.
<u>4696682</u>	September 1987	Hirai et al.
<u>4702903</u>	October 1987	Keefer
<u>4726816</u>	February 1988	Fuderer
<u>4743276</u>	May 1988	Nishida et al.
<u>4758253</u>	July 1988	Davidson et al.
<u>4759997</u>	July 1988	Ohyauchi et al.
<u>4781735</u>	November 1988	Tagawa et al.
<u>4783433</u>	November 1988	Tajima et al.
<u>4790858</u>	December 1988	Sircar
<u>4801308</u>	January 1989	Keefer
<u>4816121</u>	March 1989	Keefer
<u>4914076</u>	April 1990	Tsuji et al.
<u>4917711</u>	April 1990	Xie et al.
<u>4963339</u>	October 1990	Krishnamurthy et al.
<u>4968329</u>	November 1990	Keefer
<u>4969935</u>	November 1990	Hay
<u>4988580</u>	January 1991	Ohsaki et al.
<u>4994331</u>	February 1991	Cohen
<u>5068159</u>	November 1991	Kinoshita
<u>5079103</u>	January 1992	Schramm
<u>5082473</u>	January 1992	Keefer
<u>5096469</u>	March 1992	Keefer
<u>5096470</u>	March 1992	Krishnamurthy
<u>5126310</u>	June 1992	Golden et al.
<u>5133784</u>	July 1992	Boudet et al.
<u>5147735</u>	September 1992	Ippommatsu et al.
<u>5175061</u>	December 1992	Hildebrandt et al.
<u>5227598</u>	July 1993	Woodmansee et al.
<u>5245110</u>	September 1993	Van Dijk et al.
<u>5246676</u>	September 1993	Hay
<u>5248325</u>	September 1993	Kagimoto et al.
<u>5256172</u>	October 1993	Keefer
<u>5256174</u>	October 1993	Kai et al.
<u>5258571</u>	November 1993	Golden et al.

<u>5271916</u>	December 1993	Vanderburgh et al.
<u>5282886</u>	February 1994	Kobayashi et al.
<u>5306575</u>	April 1994	Camara et al.
<u>5328503</u>	July 1994	Kumar et al.
<u>5360679</u>	November 1994	Buswell et al.
<u>5366818</u>	November 1994	Wilkinson et al.
<u>5393326</u>	February 1995	Engler et al.
<u>5411578</u>	May 1995	Watson et al.
<u>5415748</u>	May 1995	Emiliani et al.
<u>5429665</u>	July 1995	Botich
<u>5431716</u>	July 1995	Ebbeson
<u>5434016</u>	July 1995	Benz et al.
<u>5441559</u>	August 1995	Petit et al.
<u>5487775</u>	January 1996*	LaCava et al.
<u>5509956</u>	April 1996	Opperman et al.
<u>5523326</u>	June 1996	Dandekar et al.
<u>5529763</u>	June 1996	Peng et al.
<u>5529970</u>	June 1996	Peng
<u>5531809</u>	July 1996	Golden et al.
<u>5543238</u>	August 1996	Strassner
<u>5593478</u>	January 1997	Hill et al.
<u>5604047</u>	February 1997	Bellows et al.
<u>5632807</u>	May 1997	Tomita et al.
<u>5645950</u>	July 1997	Benz et al.
<u>5646305</u>	July 1997	Wagner et al.
<u>5656067</u>	August 1997	Watson et al.
<u>5658370</u>	August 1997	Vigor et al.
<u>5711926</u>	January 1998	Knaebel
<u>5714276</u>	February 1998	Okamoto
<u>5766311</u>	June 1998	Ackley et al.
<u>5811201</u>	September 1998	Skowronski
<u>5827358</u>	October 1998	Kulish et al.
<u>5876486</u>	March 1999	Steinwandel et al.
<u>5877600</u>	March 1999	Sonntag
<u>5891217</u>	April 1999	Lemcoff et al.
<u>5900329</u>	May 1999	Reiter et al.
<u>5917136</u>	June 1999	Gaffney et al.
<u>5925322</u>	July 1999	Werth
<u>5955039</u>	September 1999	Dowdy
<u>5958109</u>	September 1999	Fuderer
<u>5968680</u>	October 1999	Wolfe et al.
<u>5980857</u>	November 1999	Kapoor et al.
<u>5981096</u>	November 1999	Hornberg et al.
<u>5998056</u>	December 1999	Divisek et al.
<u>6022399</u>	February 2000	Ertl et al.
<u>6045933</u>	April 2000	Okamoto
<u>6051050</u>	April 2000	Keefer et al.

429/23 X

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<u>6060032</u>	May 2000	Hable et al.
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<u>6077620</u>	June 2000	Pettit
<u>6090312</u>	July 2000	Ziaka et al.
<u>6143057</u>	November 2000	Bulow et al.
<u>6162558</u>	December 2000	Borup et al.
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<u>6190623</u>	February 2001	Sanger et al.
<u>6190791</u>	February 2001	Hornburg
<u>6200365</u>	March 2001	Eimer et al.
<u>6210822</u>	April 2001	Abersfelder et al.
<u>6231644</u>	May 2001	Jain et al.
<u>6255010</u>	July 2001	George et al.
<u>6280865</u>	August 2001	Eisman et al.
<u>6283723</u>	September 2001	Milburn et al.
<u>6293998</u>	September 2001	Dolan et al.
<u>6296823</u>	October 2001	Ertl et al.
<u>6312843</u>	November 2001	Kimbara et al.
<u>6358300</u>	March 2002	Fornof et al.
<u>6398853</u>	June 2002	Keefer et al.
<u>6406523</u>	June 2002	Connor et al.
<u>6428915</u>	August 2002	Ban et al.
<u>6492048</u>	December 2002	Draper et al.
<u>6607854</u>	August 2003	Rehg et al.
<u>6667128</u>	December 2003	Edlund
<u>6692545</u>	February 2004	Gittleman et al.
<u>2001/0047824</u>	December 2001	Hill et al.
<u>2002/0004157</u>	January 2002	Keefer et al.
<u>2002/0098394</u>	July 2002	Keefer et al.
<u>2002/0104518</u>	August 2002	Keefer et al.
<u>2002/0110503</u>	August 2002	Gittleman et al.
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<u>2002/0112479</u>	August 2002	Keefer et al.
<u>2002/0127442</u>	September 2002	Connor et al.
<u>2002/0142198</u>	October 2002	Towler et al.
<u>2002/0142208</u>	October 2002	Keefer et al.
<u>2003/0143448</u>	July 2003	Keefer et al.
<u>2003/0157390</u>	August 2003	Keefer et al.
<u>2004/0005492</u>	January 2004	Keefer et al.

FOREIGN PATENT DOCUMENTS

FOREIGN-PAT-NO	PUBN-DATE	COUNTRY	CLASS
1256038	June 1989	CA	
2016045	August 1994	CA	

2109055	February 1999	CA
2087972	January 2000	CA
2087973	January 2001	CA
3913 581	November 1990	DE
0 341 189	August 1989	EP
0 345 908	December 1989	EP
0 143 537	March 1990	EP
0 143 537	March 1990	EP
0 681 860	July 1996	EP
0 691 701	October 1996	EP
0 737 648	October 1996	EP
0 750 361	December 1996	EP
0 751 045	January 1997	EP
0 853 967	July 1998	EP
1 095 689	October 1999	EP
1 070 531	January 2001	EP
1 172 772	January 2002	EP
2 042 365	September 1980	GB
59075574	April 1984	JP
62-274561	November 1987	JP
62 274561	November 1987	JP
62 278770	December 1987	JP
63 166137	July 1988	JP
63 228572	September 1988	JP
04 206161	July 1992	JP
05 166528	July 1993	JP
07094200	July 1995	JP
80 45526	February 1996	JP
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10 027621	January 1998	JP
10 325360	December 1998	JP
10325360	December 1998	JP
11214021	August 1999	JP
WO 94/04249	August 1992	WO
WO 96/13871	May 1996	WO
WO 98/29182	September 1998	WO
WO 99/01202	January 1999	WO
WO 99/19249	April 1999	WO
WO 99/28013	June 1999	WO
WO 99/46032	September 1999	WO
WO 00/16425	March 2000	WO
WO 00/16880	March 2000	WO
WO 00/76630	December 2000	WO
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WO 02/24309	March 2002	WO
WO 02/35623	May 2002	WO
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Hufton et al., "Sorption Enhanced Reaction Process for Hydrogen Production," AIChE Journal, vol. 45 No. 2, pp. 248-256 (Feb. 1999). cited by other
Carvill et al., AIChE J. 42(10):2765-2772, 1996. cited by other
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Iyuke et al., Chemical Engineering Science 55:4745-4755, 2000. cited by other
International Search Report from International Application No. PCT/CA99/00823. cited by other
International Search Report from International Application No. PCT/CA02/00368. cited by other

ART-UNIT: 1745

PRIMARY-EXAMINER: Kalafut; Stephen J.

ATTY-AGENT-FIRM: Klarquist Sparkman, LLP

ABSTRACT:

Disclosed is a high temperature fuel cell power generation system that includes a high temperature fuel cell having an anode inlet and exhaust, and a cathode inlet and exhaust. The system also includes a gas separation means operable to recover hydrogen gas from the anode exhaust and to provide at least a portion of such hydrogen gas for recycle to the anode inlet. The system further includes energy recovery means operable to recover energy from the fuel cell exhaust gases and to provide at least a portion of such recovered energy to drive mechanical loads associated with the operation of the gas separation means, wherein a portion of the recovered hydrogen gas is provided for export from the generation system as hydrogen fuel.

30 Claims, 17 Drawing figures

Full	Title	Citation	Front	Review	Classification	Date	Reference	Searched	Attachments	Claims	KWIC	Draft D
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 5. Document ID: US 6525104 B2

L4: Entry 5 of 5

File: USPT

Feb 25, 2003

US-PAT-NO: 6525104

DOCUMENT-IDENTIFIER: US 6525104 B2

** See image for Certificate of Correction **

TITLE: Steam reforming

DATE-ISSUED: February 25, 2003

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Abbott; Peter Edward James	Eaglescliffe			GB

ASSIGNEE-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY	TYPE CODE
Johnson Matthey PLC				GB	03

APPL-NO: 09/781312 [PALM]

DATE FILED: February 13, 2001

PARENT-CASE:

This application is a continuation of PCT/GB99/02286 filed Jul. 15, 1999.

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY	APPL-NO	APPL-DATE
GB	9817526	August 13, 1998

INT-CL-ISSUED: [07] C07C 27/06

INT-CL-CURRENT:

TYPE IPC	DATE
CIPS <u>B01 J 19/00</u>	20060101
CIPS <u>C10 G 2/00</u>	20060101
CIPS <u>B01 J 8/04</u>	20060101
CIPS <u>B01 J 8/02</u>	20060101
CIPS <u>B01 J 8/06</u>	20060101
CIPS <u>C01 B 3/38</u>	20060101
CIPS <u>C01 B 3/00</u>	20060101

US-CL-ISSUED: 518/704; 252/373

US-CL-CURRENT: 518/704; 252/373

FIELD-OF-CLASSIFICATION-SEARCH: 252/373, 518/704

See application file for complete search history.

PRIOR-ART-DISCLOSED:

U.S. PATENT DOCUMENTS

PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<u>2579843</u>	December 1951	Mader	
<u>4479925</u>	October 1984	Shires	
<u>4822521</u>	April 1989	Fuderer	252/373
<u>4824658</u>	April 1989	Karafian	
<u>4910228</u>	March 1990	Lywood	
<u>5300275</u>	April 1994	Lywood	252/373

5855815

January 1999

Park et al.

252/373

FOREIGN PATENT DOCUMENTS

FOREIGN-PAT-NO	PUBN-DATE	COUNTRY	CLASS
0342610	November 1989	EP	
2179366	August 1985	GB	

ART-UNIT: 1754

PRIMARY-EXAMINER: Langel; Wayne A.

ATTY-AGENT-FIRM: Pillsbury Winthrop LLP

ABSTRACT:

Production of synthesis gas for use for synthesising carbon-containing compounds, typically having a hydrogen to carbon monoxide molar ratio of about 2 and a low carbon dioxide content, by primary reforming a gaseous mixture containing hydrocarbons, 0.6 to 2 moles of steam per gram atom of hydrocarbon and 0.2 to 0.6 moles of recycled carbon dioxide per gram atom of hydrocarbon, in a heat exchange reformer and then secondary reforming the resultant primary reformed gas, heating the heat exchange reformer with the resultant secondary reformed gas; cooling and condensing steam from the secondary reformed gas to give a de-watered gas stream having a carbon dioxide content below 20% by volume. The recycled carbon dioxide is recovered from the de-watered gas stream, before or after use thereof for the synthesis reaction.

10 Claims, 3 Drawing figures

Full	Title	Citation	Front	Review	Classification	Date	Reference	Claims	KWIC	Drawn D
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Clear	Generate Collection	Print	Fwd Refs	Bkwd Refs	Generate OACS
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Term	Documents
CARBON	1699737
CARBONS	114677
DIOXIDE	595979
DIOXIDES	8186
SEPARAT\$3	0
SEPARAT	25374
SEPARATA	450
SEPARATAA	2
SEPARATAB	1

SEPARATABE	2
SEPARATABL	2
(L2 AND SEPARAT\$3 NEAR4 CARBON DIOXIDE).PGPB,USPT,USOC,EPAB,JPAB,DWPI.	5

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